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mole ratio. excess reactant. limiting
reactant. The study of quantitative
relationships between the amounts of....
In a balanced equation, the ratio
between the numbers of moles.... A
reactant that remains after a chemical
reaction stops.

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barometer. standard atmospheric
pressure. manometer. dalton's law of
partial pressures. a device that
measures atmospheric pressure. It is
filled with.... $1 \text{ atm} = 760 \text{ mm Hg}$...
 $\text{pascal} = 1 \text{ N/m}^2$... $1 \text{ atm} = 101.3$...
 $\text{kPa} = 101,325 \text{ Pa}$. measures gases in a

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closed container.

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barometer. standard atmospheric

pressure. manometer. daltons law of

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filled with.... 1 atm = 760 mm Hg...
pascal = 1N/m^2 ... 1 atm = 101.3...
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Molality moles of solute L of solution
moles of solute kg of solvent Percent
mass grams of solute $\times 100$ grams of
solution Mole Fraction mol solute total

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moles Density of Solution grams solution
mL solution X Solution Formation
Electrolytic ions present in solution
Hydrated ions. interactions keep the ions
from recrystallizing Not everything that
dissolves in water is an electrolytic
solution. o Sugar dissolves.

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