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14 Acids and Bases
CHAPTER 14 REVIEW . Acids and Bases. SHORT ANSWER Answer the following questions in the space provided. 1. a. Write the two equations that show the two-stage ionization of sulfurous acid in water. stage 1: H2SO3(aq) + H2O(l) → H3O+(aq) + HSO3-(aq) b. Which stage of ionization usually produces more ions? Explain your answer.

CHAPTER 14 REVIEW Acids and Bases
CHAPTER 14 REVIEW Ions in Aqueous Solutions and Colligative Properties SECTION 14-1 SHORT ANSWER Answer the following questions in the space provided.

14 Ions in Aqueous Solutions and Colligative Properties
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CHAPTER 16 REVIEW Reaction Energy MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Describe Hess's law. The overall enthalpy change in a reaction is equal to the sum of enthalpy changes for the individual steps in the process. 2. What determines the amount of energy absorbed by a material when it is heated?

16 Reaction Energy
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